

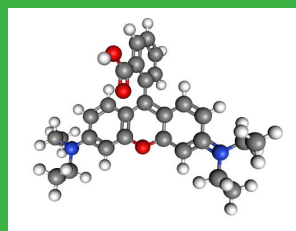


ANALYSIS OF THE PROBLEM

Water is the most precious resource for any nation today. The possibility of having enough of it, with the right purity for the type of use (agriculture or human consumption), is the main objective. In our area, surface water is mainly contaminated with traces of pesticides and heavy metals. The project was born from the need to a path towards environmental sustainability by reducing waste and promoting the reuse of water.

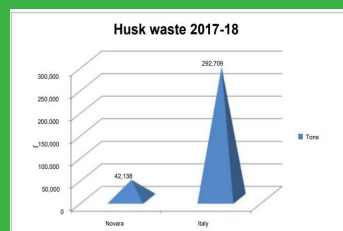
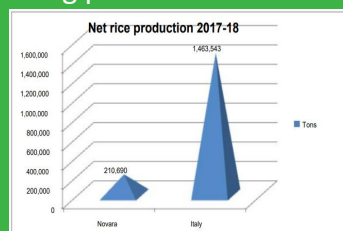
POLLUTION FROM DYES

Pollution from dye effluents has become a major environmental problem in the last decade due to the increasing use of dyes in various applications., the global textile industry is the main source of these effluents. The dye that we analysed is Rhodamine B, a dye belongs to the class of xanthenes.

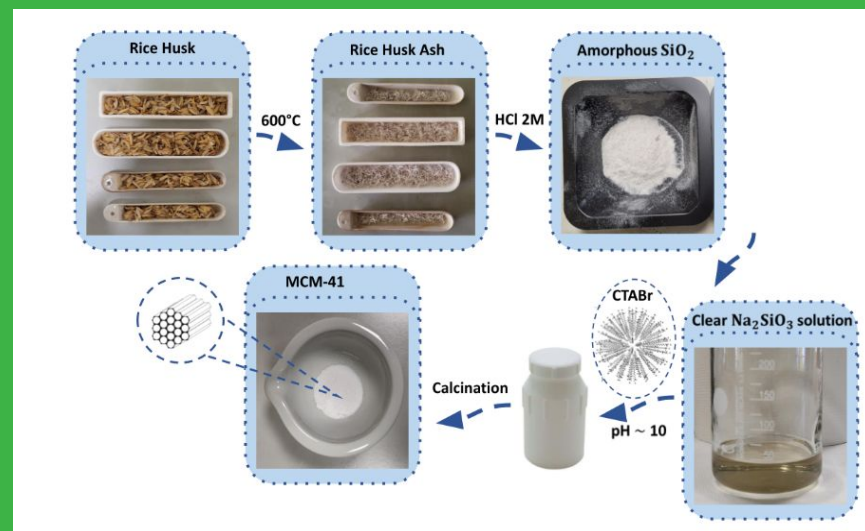


OUR PROJECT

The Rawtrap project was born from the need to address the important issue of water purification, using waste materials from rice production. Specifically, it focuses on the use of rice husk, named **CHAFF**, a waste material rich in silica. This allows us to synthesize nanoparticles called MCM-41 which are molecular sieve capable of capturing pollutants.

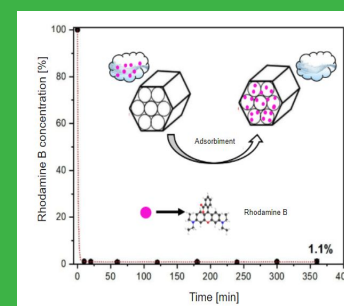
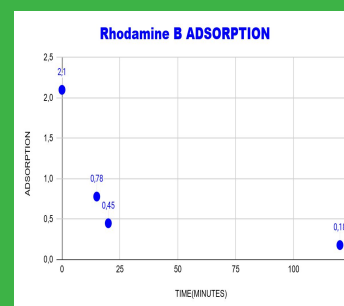
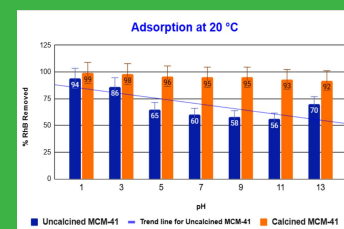
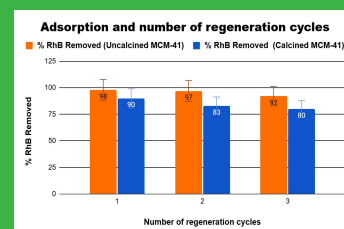


EXPERIMENTAL PROCEDURE

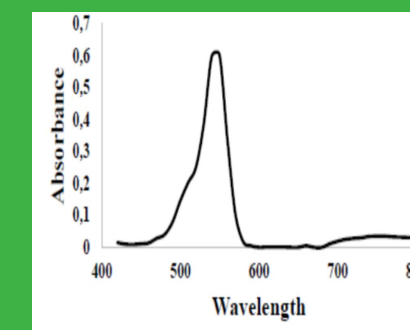
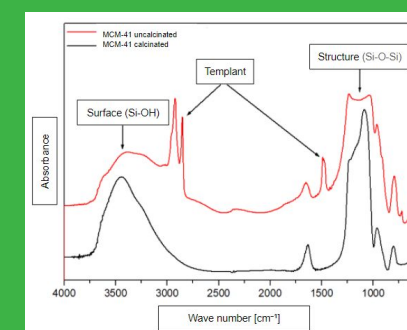
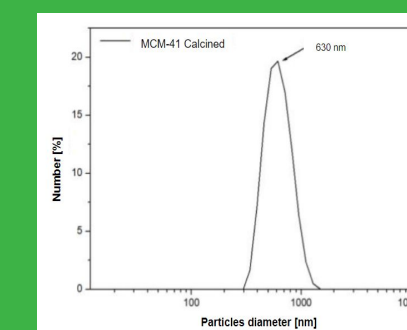
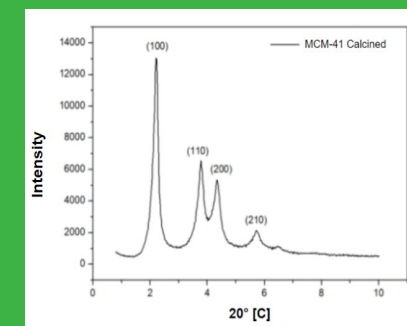


POLLUTANTS CAPTURE TESTS

The experiments were carried out (at room temperature) by dissolving 100 [mg] of MCM-41 in 20 [ml] of a Rhodamine B $2 \cdot 10^{-1}$ [mmol·L⁻¹] solution. The pH of the dye solution is an important parameter in determining the adsorption capacity.



ANALYSIS: EDX, FTIR, DLS, UV-VIS



CONCLUSIONS

Based on the results obtained, it can be concluded that the production of absorbent materials such as MCM-41 according to the principles of Green Chemistry, to be used for purifying wastewater from the textile and dyeing industry, is promising. The best results were obtained with samples of MCM-41 calcined at 700°C.